

## Physiology at a Glance

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# Physiology at a Glance

### Fourth Edition

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### **Preface**



hysiology is defined as 'the scientific study of the bodily function of living organisms and their parts'. There is a natural symbiosis between function (physiology) and structure (anatomy) from which physiology emerged as a separate discipline in the late 19th century. A good understanding of anatomy and physiology is an essential prerequisite for understanding what happens when things go wrong - the structural abnormalities and pathophysiology of disease - and as such underpins all biomedical studies and medicine itself. Following a century of reductionism, where the focus of research has progressively narrowed down to the function of individual proteins and genes, there is now a resurgence in integrative physiology, as it has been realized that to make sense of developments such as the Human Genome Project we have to understand body function as an integrated whole. This is considerably more complex than just the sum of its parts because of the multiplicity of interactions involved. True understanding of the role of a single gene, for example, can only be gained when placed in the context of the whole animal, as reflected by the often unexpected effects of knock-out of single genes on the phenotype of mice.

This volume is designed as a concise guide and revision aid to core topics in physiology, and should be useful to all students following a first-year physiology course, whether they are studying single honours, biomedical sciences, nursing, medicine or dentistry. It should also be useful to those studying systembased curricula. The layout of *Physiology at a Glance* follows that of the other volumes in the *At a Glance series*, with a two-page spread for each topic (loosely corresponding to a lecture),

comprising a large diagram on one page and concise explanatory text on the other. For this fourth edition we have extensively revised the text and figures, there are three completely new chapters, on Cell signalling, Thermoregulation, and Altitude and aerospace physiology, and we have added a Glossary.

Physiology is a large subject, and in a book this size we cannot hope to cover anything but the core and basics. *Physiology at a Glance* should therefore be used primarily to assist basic understanding of key concepts and as an assistance to revision. Deeper knowledge should be gained by reference to full physiology and system textbooks, and in third-year honours programmes to original peer-reviewed papers. Students may find one or two sections of this book difficult, such as that on the physics of flow and diffusion, and detailed elements of cell signalling. Though such material may not be included in some introductory physiology courses, an understanding of these concepts can assist in learning how body systems behave in the way they do, and in understanding primary research papers.

In revising this fourth edition we have been helped immensely by constructive criticism and suggestions from our colleagues and students, and junior and senior reviewers of the last edition. We thank all those who have given us such advice; any errors are ours and not theirs. We would also like to thank the team at Wiley-Blackwell who provided great encouragement and support throughout the project.

Jeremy Ward Roger Linden



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Mehta, A. and Hoffbrand, V. (2009) *Haematology at a Glance* (3rd edition). Wiley-Blackwell, Oxford.

## **Abbreviations**



1,25-(OH)<sub>2</sub>D 1,25-dihydroxycholecalciferol 2.3-DPG 2,3-diphosphoglycerate 5-HT 5-hydroxytryptamine; serotonin μG micro-gravity (weightlessness)

**ACE** angiotensin-converting enzyme

**ACh** acetylcholine

**ACTH** adrenocorticotrophic hormone ADH antidiuretic hormone (also called

vasopressin)

**ADP** adenosine diphosphate

**AIDS** acquired immune deficiency syndrome

anti-Müllerian hormone AMH **AMS** acute mountain sickness **ANP** atrial natriuretic peptide **ANS** autonomic nervous system

AP action potential

**APC** active protein C or antigen

presenting cell

**ATP** adenosine triphosphate **ATPase** enzyme that splits ATP **AV** node atrioventricular node (heart) **AVAs** arteriovenous anastomoses

BAT brown adipose tissue **BSA** body surface area

**BTPS** body temperature and pressure,

saturated with water

calmodulin CaM

calcium-calmodulin kinase CaM-kinase

**cAMP** cyclic adenosine monophosphate **CaSR** calcium-sensing receptor (protein)

CCK cholecystokinin

CDI central diabetes insipidus

cyclic guanosine monophosphate **cGMP** 

CICR Ca<sup>2+</sup>-induced Ca<sup>2+</sup> release CNS central nervous system

CO cardiac output

COMT catechol-O-methyl transferase

COX cyclooxygenase

**CRH** corticotrophin-releasing hormone

**CSF** cerebrospinal fluid **CVP** central venous pressure

Da Dalton (unit for molecular weight)

DAG diacylglycerol

**DHEA** dehydroepiandrosterone

O<sub>2</sub> diffusing capacity in lung; transfer  $D_{LO_2}$ 

factor

DNA deoxyribonucleic acid **DOPA** dihydroxyphenylalanine

equilibrium potential for ion (e.g. K+, E<sub>(ion)</sub>

Na<sup>+</sup>. Ca<sup>2+</sup> or Cl<sup>-</sup>)

**ECF** extracellular fluid ECG (EKG) electrocardiogram (or graph)

**EDP** end diastolic pressure **EDV** end diastolic volume **EGF** epidermal growth factor  $E_{m}$ membrane potential **EMG** electromvogram **EPO** erythropoietin **EPP** end plate potential **ERV** expiratory reserve volume **ESV** end systolic volume **ETC** electron transport chain

F<sub>ab</sub> hypervariable region of antibody

molecule

constant region of antibody molecule FEV₁ forced expiratory volume in 1 s

**FFA** free fatty acids

**FRC** 

**FSH** 

**HAPE** 

**FGF** fibroblast growth factor

F<sub>N2</sub> (F<sub>O2</sub>) fractional concentration of nitrogen

> (oxygen) in a gas mixture functional residual capacity follicle-stimulating hormone

**FVC** forced vital capacity

G-LOC G-forces induced loss of

consciousness

G-protein GTP-binding protein **GDP** guanosine diphosphate **GFR** glomerular filtration rate

GH growth hormone

**GHRH** growth hormone-releasing hormone

GI gastrointestinal

**GIP** gastric inhibitory peptide GLP-1 glucagon-like peptide 1 GLUT-1, 2 or 4 glucose transporters

**GnRH** gonadotrophin-releasing hormone **GPCR** G-protein-coupled receptor **GRP** gastrin-releasing peptide guanosine triphosphate **GTP GTPase** enzyme that splits GTP HACE high-altitude cerebral oedema

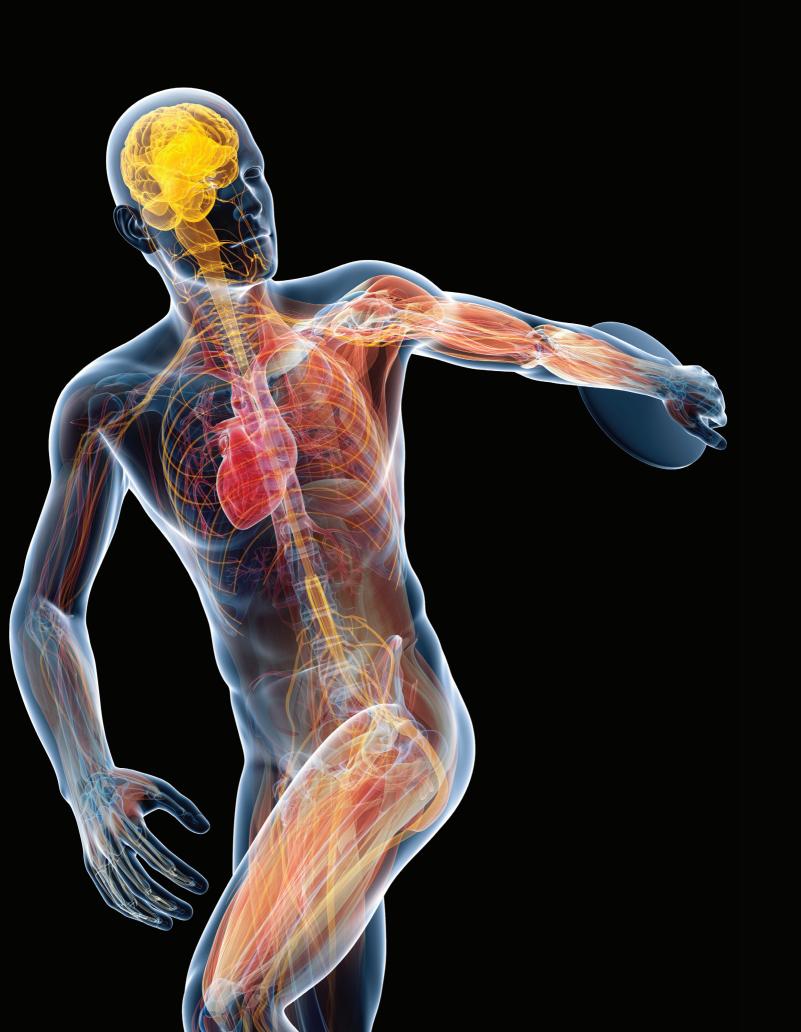
high-altitude pulmonary oedema haemoglobin concentration [Hb]

HbA adult haemoglobin **PLC** phospholipase C **PMCA** plasma membrane Ca2+ ATPase **HbF** fetal haemoglobin **hCG** human chorionic gonadotrophin partial pressure of oxygen Pos HIV human immunodeficiency virus **PRR** pattern recognition receptor **HMWK** high molecular weight kininogen **PTH** parathyroid hormone **ICF** intracellular fluid Ras, Rho small monomeric GTPases IgA, E, G, M immunoglobulin A, E, G or M **ROC** receptor-operated channels IGF-1 or 2 insulin-like growth factor (1 or 2) **ROMK** renal outer medullary potassium channel IL-1b or 6 interleukin-1β or 6 **RPF** renal plasma flow inositol trisphosphate **RTK** receptor tyrosine kinase  $IP_3$ IRS-1 insulin receptor substrate 1 RV residual volume or right ventricle SA node **IRV** inspiratory reserve volume sinoatrial node smooth endoplasmic reticulum Ca2+ **ISF** interstitial fluid **SERCA JAK** Janus kinase **ATPase JGA** juxtaglomerular apparatus SH<sub>2</sub> Src-homology 2 intracellular protein associated with LH luteinizing hormone **SMAD** MAO monoamine oxidase streptokinases SOC MAP mean arterial pressure store-operated channels MAPK(K) mitogen-activated protein kinase SP Substance P sarcoplasmic reticulum (kinase) SR **MEPP** miniature end plate potentials Src a non-receptor tyrosine kinase MHC I. II major histocompatibility complex I or II SST somatostatin signal transduction and activation of MIH melanotrophin-inhibiting hormone **STAT MLC** myosin light chain transcription (protein) **MLCK** myosin light chain kinase **STIM** stromal interaction molecule **MLCP** myosin light chain phosphatase **STPD** standard temperature and pressure, **mRNA** messenger RNA dry gas melanotrophin-stimulating hormone SV stroke volume **MSH SWVP** Na<sup>+</sup> pump Na+-K+ ATPase saturated water vapour pressure NAD+ or (NADH) nicotinic adenine dinucleotide mono- or di-iodotyrosine T<sub>1 or 2</sub> (oxidized and reduced forms)  $T_3$ tri-iodothyronine **NCX** Na+-Ca<sup>2+</sup> exchanger T₄ thyroxine NDI nephrogenic diabetes insipidus Core temperature  $T_{C}$ NGF nerve growth factor TF tissue factor TGFβ NK natural killer (cells) transforming growth factor β **NMJ** neuromuscular junction TH thyroid hormone NO nitric oxide **TLC** total lung capacity NOS nitric oxide synthase tubular transport maximum (kidney) T<sub>m</sub> P2Y or P2X purinergic receptor type 2Y or 2X **TNF** tumour necrosis factor PAH para-aminohippuric acid **TNZ** thermoneutral zone **PAMP** pathogen-associated molecular **tPA** tissue plasminogen activator pattern **TPR** total peripheral resistance  $P_B$ barometric pressure TRa thyroid hormone receptor **PDGF** platelet-derived growth factor **TRE** thyroid response element peak expiratory flow rate transfer RNA **PEFR tRNA** thyroid-stimulating hormone PGE<sub>2</sub> prostaglandin E2 **TSH TUC** time of useful consciousness PGI<sub>2</sub> prostacyclin (prostaglandin l<sub>2</sub>) PI-3 kinase phosphatidylinositol-3 kinase TV tidal volume pK negative log of dissociation constant TXA<sub>2</sub> thromboxane A<sub>2</sub> (buffers) UCP-1, 2 or 3 uncoupling protein-1, 2 or 3 **PKA** protein kinase A V<sub>△</sub>/Q mismatch ventilation—perfusion mismatch (lungs) **PKC** protein kinase C VC vital capacity **PKG** protein kinase G **VIP** vasoactive intestinal polypeptide PLA<sub>2</sub> phospholipase A2 vWF von Willebrand factor

## About the companion website







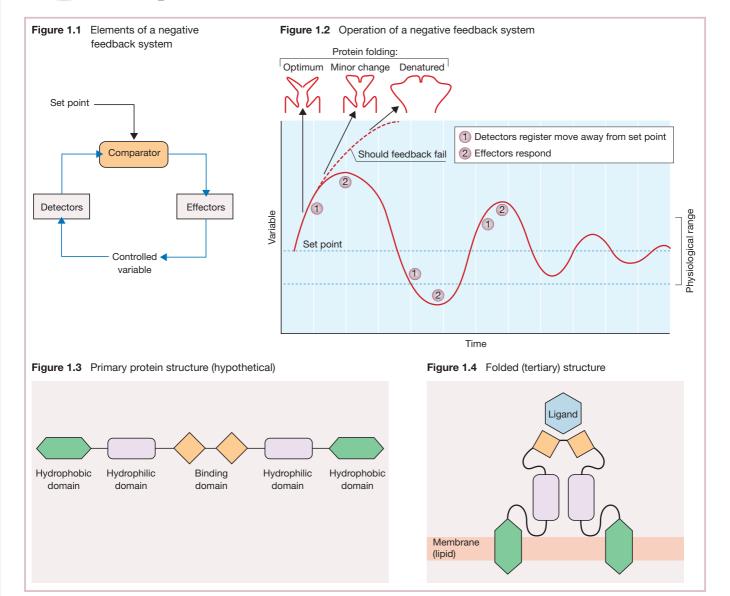
## Introduction



#### **Chapters**

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## Homeostasis and the physiology of proteins



laude Bernard (1813–1878) first described 'le mileau intérieur' and observed that the internal environment of the body remained remarkably constant (or in equilibrium) despite the ever changing external environment. The term homeostasis was not used until 1929 when Walter Cannon first used it to describe this ability of physiological systems to maintain conditions within the body in a relatively constant state of equilibrium. It is arguably the most important concept in physiology.

**Homeostasis** is Greek for 'staying the same'. However, this so-called **equilibrium** is not an unchanging state but is a dynamic state of equilibrium causing a **dynamic constancy** of the internal environment. This **dynamic constancy** arises from the variable responses caused by changes in the external environment. Homeostasis maintains most physiological systems

and examples are seen throughout this book. The way in which the body maintains the  $\mathrm{H}^+$  ion concentration of body fluids within narrow limits, the control of blood glucose by the release of insulin, and the control of body temperature, heart rate and blood pressure are all examples of homeostasis. The human body has literally thousands of control systems. The most intricate are genetic control systems that operate in all cells to control intracellular function as well as all extracellular functions. Many others operate within organs to control their function; others operate throughout the body to control interaction between organs. As long as conditions are maintained within the normal physiological range within the internal environment, the cells of the body continue to live and function properly. Each cell benefits from homeostasis and in turn, each cell contributes its share towards the maintenance of homeostasis. This reciprocal

interplay provides continuity of life until one or more functional systems lose their ability to contribute their share. Moderate dysfunction of homeostasis leads to sickness and disease, and extreme dysfunction of homeostasis leads to death.

#### **Negative feedback control**

Most physiological control mechanisms have a common basic structure. The factor that is being controlled is called the vari**able.** Homeostatic mechanisms provide the tight regulation of all physiological variables and the most common type of regulation is by *negative* feedback. A negative feedback system (Figure 1.1) comprises: detectors (often neural receptor cells) to measure the variable in question; a **comparator** (usually a neural assembly in the central nervous system) to receive input from the detectors and compare the size of the signal against the desired level of the variable (the set point); and effectors (muscular and/or glandular tissue) that are activated by the comparator to restore the variable to its set point. The term 'negative feedback' comes from the fact that the effectors always act to move the variable in the opposite direction to the change that was originally detected. Thus, when the partial pressure of CO<sub>2</sub> in blood increases above 5.3 kPa (40 mmHg), brain stem mechanisms increase the rate of ventilation to clear the excess gas, and vice versa when CO<sub>2</sub> levels fall below 5.3 kPa (Chapter 32). The term 'set point' implies that there is a single optimum value for each physiological variable; however, there is some tolerance in all physiological systems and the set point is actually a narrow range of values within which physiological processes will work normally (Figure 1.2). Not only is the set point not a point, but it can be reset in some systems according to physiological requirements. For instance, at high altitude, the low partial pressure of  $O_2$  in inspired air causes the ventilation rate to increase. Initially, this effect is limited due to the loss of CO<sub>2</sub>, but, after 2–3 days, the brain stem lowers the set point for CO2 and allows ventilation to increase further, a process known as **acclimatization** (Chapter 14).

A common operational feature of all negative feedback systems is that they induce oscillations in the variable that they control (Figure 1.2). The reason for this is that it takes time for a system to detect and respond to a change in a variable. This delay means that feedback control always causes the variable to overshoot the set point slightly, activating the opposite restorative mechanism to induce a smaller overshoot in that direction, until the oscillations fall within the range of values that are optimal for physiological function. Normally, such oscillations have little visible effect. However, if unusually long delays are introduced into a system, the oscillations can become extreme. Patients with congestive heart failure sometimes show a condition known as Cheyne-Stokes' breathing, in which the patient undergoes periods of deep breathing interspersed with periods of no breathing at all (apnoea). This is partly due to the slow flow of blood from the lungs to the brain, which causes a large delay in the detection of blood levels of  $CO_2$ .

Some physiological responses use *positive* feedback, causing rapid amplification. Examples include initiation of an action potential, where sodium entry causes depolarization which further increases sodium entry and thus more depolarization

(Chapter 5), and certain hormonal changes, particularly in reproduction (Chapter 53). Positive feedback is inherently unstable, and requires some mechanism to break the feedback loop and stop the process (an off switch), such as time-dependent inactivation of sodium channels in the first example and the birth of the child in the second.

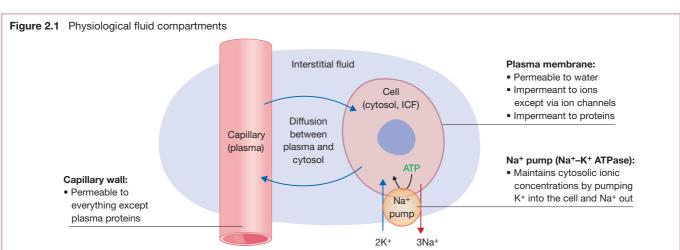
## Protein form and function are protected by homeostatic mechanisms

The homeostatic mechanisms that are described in detail throughout this book have evolved to protect the integrity of the protein products of gene translation. Normal functioning of proteins is essential for life, and usually requires binding to other molecules, including other proteins. The specificity of this binding is determined by the three-dimensional shape of the protein. The primary structure of a protein is determined by the sequence of amino acids (Figure 1.3). Genetic mutations that alter this sequence can have profound effects on the functionality of the final molecule. Such gene polymorphisms are the basis of many genetically based disorders. The final shape of the molecule (the tertiary structure), however, results from a process of **folding** of the amino acid chain (Figure 1.4). Folding is a complex process by which a protein achieves its lowest energy conformation. It is determined by electrochemical interactions between amino acid side-chains (e.g. hydrogen bonds, van der Waals' forces), and is so vital that it is overseen by molecular **chaperones**, such as the **heat shock proteins**, which provide a quiet space within which the protein acquires its final shape. In healthy tissue, cells can detect and destroy misfolded proteins, the accumulation of which damages cells and is responsible for various pathological conditions, including Alzheimer's disease and Creutzfeldt-Jakob disease. Folding ensures that the functional sequences of amino acids (domains) that form, e.g. binding sites for other molecules or hydrophobic segments for insertion into a membrane, are properly orientated to allow the protein to serve its function.

The relatively weak nature of the forces that cause folding renders them sensitive to changes in the environment surrounding the protein. Thus, alterations in acidity, osmotic potential, concentrations of specific molecules/ions, temperature or even hydrostatic pressure can modify the tertiary shape of a protein and change its interactions with other molecules. These modifications are usually reversible and are exploited by some proteins to detect alterations in the internal or external environments. For instance, nerve cells that respond to changes in CO<sub>2</sub> (chemoreceptors; Chapter 32) possess ion channel proteins (Chapter 4) that open or close to generate electrical signals (Chapter 5) when the acidity of the medium surrounding the receptor (CO<sub>2</sub> forms an acid in solution) alters by more than a certain amount. However, there are limits to the degree of fluctuation in the internal environment that can be tolerated by proteins before their shape alters so much that they become non-functional or irreversibly damaged, a process known as denaturation (this is what happens to egg-white proteins in cooking). Homeostatic systems prevent such conditions from arising within the body, and thus preserve protein functionality.



## **Body water compartments and** physiological fluids



| Constituents of physiological fluids (approximate values, intracellular varies between tissues) |  |        |              |               |                            |  |  |
|---|--|--------|--------------|---------------|----------------------------|--|--|
|   |  | Plasma | Interstitial | Intracellular | Unit                       |  |  |
| Water:  | % total body water                               | 13%    | 22%          | 65%           | %                          |  |  |
|   | (volume in a 70 kg person)                       | (3.5)  | (9.5)        | (27)          | L                          |  |  |
| Osmolality  |  | 290    | 290          | 290           | mosmol/kg H <sub>2</sub> O |  |  |
| Cations:  | Na+  | 140    | 140          | 10            | mmol/L                     |  |  |
|   | K <sup>+</sup>                                   | 4      | 4            | 140           | mmol/L                     |  |  |
|   | Ca <sup>2+</sup> (free)                          | 1      | 1            | 0.0001        | mmol/L                     |  |  |
| Anions:   | CI-  | 108    | 129          | 3–30          | mmol/L                     |  |  |
|   | HCO <sub>3</sub> -                               | 26     | 26           | 9             | mmol/L                     |  |  |
|   | Proteins-  | 10     | 1            | 50            | mmol/L                     |  |  |
|   | Other anions (mainly $PO_4^{3-}$ , $SO_4^{3-}$ ) | 3      | 0            | 60–88         | mmol/L                     |  |  |

Notes: Ca<sup>2+</sup> (and Mg<sup>2+</sup>) tend to bind to plasma proteins, and their free concentrations are about 50% of the total. Ionic concentrations are sometimes given in mEq/L to reflect the amount of charge, where an equivalent (Eq) is 1 mole of charge. So 1 Eq of a monovalent ion such as  $Na^+ = 1$  mole, but 1 Eq of  $Ca^{2+} = 0.5$  mole

Figure 2.2 Effects of ingesting fluids of differing osmotic potential

