

Vitamin B₁₂ and B₁₂-Proteins

Edited by

Bernhard Kräutler, Duilio Arigoni and
Bernard T. Golding

Lectures presented at the
4th European Symposium
on Vitamin B₁₂ and B₁₂-Proteins

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The cover picture shows a cartoon of B₁₂-dependent methionine synthase (see contribution by Drennan et al. in this book).

The picture was kindly provided by Martin Tollinger, University of Innsbruck.

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Foreword

In an article written in the seventies, the decade of the previous B₁₂ symposium, I was referring metaphorically to the endeavour of a chemical synthesis of vitamin B₁₂ as an attempt to "connect X-ray island with the mainland of chemical experience". Marching step by step towards this molecular structure by chemical synthesis was meant to provide us with all that knowledge of chemical properties of the complex molecule, the X-ray structure determination of which had withheld from us. Of course, achieving a chemical synthesis of this vitamin was never expected to open a chemical route for B₁₂ production - it was the extension of the frontiers of natural product synthesis of that time that was at stake.

It is true, however, that any new molecular territory discovered by the recognition of important practical consequences of its function and then chartered structurally by physical methods for further study, constitutes a challenge to the mind of the scientist the conqueror, the *homo faber*, the one who is archetypically driven to put his foot on any important newly discovered molecular territory by occupying it, that is, by making it in order to possess and explore it. This is the drive that reflects itself in the physicist's Richard Feynman dictum "What I cannot create, I do not understand" and, moreover, it is that human compulsion that lies at the heart of the inextricable interrelation between science and technology.

With the proceedings of the B₁₂-symposium of the nineties now in hand, I cannot help but be fascinated by the accounts of my colleagues summarizing their momentous achievements of the research on the biosynthesis of vitamin B₁₂ during the last decade. It becomes clear to me that the metaphor referring to the conquest of "X-ray island B₁₂" has acquired a radically expanded meaning, one which could hardly have been considered before, certainly not within the bioorganic and natural product chemistry of the time. What I mean is the breathtaking potential for discovery and learning in the realm of biomolecular architecture brought about by the advent of molecular genetics, the rigorous new dimension of progress which is made possible by the judicious application of these (in the earthbound chemist's eye miraculous) methods in biosynthesis research, and, as a consequence of these developments, the exciting prospect of a totally enzymic *in vitro* synthesis of complex biomolecules, a new kind of natural product total synthesis altogether.

Vitamin B₁₂, this exceptional biomolecule with its exquisite molecular complexity, once again is paving the way towards progress by challenging and by leading researchers to move forward methodologically as well as conceptually to unprecedented levels of inquiry and experimentation. B₁₂ has fulfilled this function before, with regard to its structure determination and its chemical total synthesis, and is doing it again in ongoing research on the intricate mechanistic pathways of B₁₂ metabolic functions. How it has exerted this influence in the task of charting the pathway(s) of its biosynthesis is truly spectacular. That story clearly will remain one of the most extraordinary chapters in the history of the life scientist's struggle in this century to discover how life, a "chemical" life after all, makes its vital molecules.

There might even be more to it. The B₁₂ molecule's exceptional structural and functional complexity, legendary as it has become to any chemist or biologist who had to deal with it, is

conjectured to contain information that refers not only to how life is operating today, but that may also bear on some aspects of life's early history. Once decoded, that information may well become part of a mosaic of circumstantial evidence that leads us to a chemical retrodiction of the evolution of some of the basic metabolic processes that we witness operating today. Such expectations are nourished and encouraged by what those recurringly dramatic discoveries on B₁₂ biosynthesis - in conjunction with what has been learned about the chemistry of the type of structures involved - reveal to us. Nature evolved not one singular route to the corrin system; interestingly enough, another one which diverges relevantly from the first has been documented and, perhaps, we should not be taken by surprise if it turned out that there are more. In sharp contrast to the structural bottleneck of the conversion of aminolaevulinic acid to uroporphyrinogen, in evolving the transformation of the latter to a corrin, Nature could select from an entire library of opportunities, a virtual library, so to say, of thermodynamically and kinetically feasible transformation pathways covering a remarkably broad spectrum of structural diversity. Nature seems to have made use of that diversity when B₁₂-producing organisms were confronted with the photosynthesis-induced oxygen-crisis billion of years ago. It is to be expected that uncovering the entire spectrum of biosynthetic pathways that may still exist today in corrinoid producing anaerobic and aerobic microorganisms would teach us an extraordinary lesson on molecular evolution of biosynthetic pathways in earliest life. That is what makes vitamin B₁₂ so fascinating, so incomparable, among the many other low molecular weight natural products, molecules which too are "chemically very interesting" and which too are "biologically of prime importance", yet which are "modern" in evolutionary terms. The B₁₂-structure with all its beautiful complexity is the architecturally richest, biosynthetically most elaborate and, therefore, the etiologically perhaps most informative member within the exquisite group of cofactor molecules which, according to the chemist's reasoning, reflect archaic types of molecular structure and are conjectured to have been part of the very beginning of metabolism as we know it today. Vitamin B₁₂ and some of those other cofactors are not only vitamins to us today, but it also looks as though their structure types had been "vitamins" - in the most direct sense of the term- to life itself in one of the critical phases of its emergence.

Needless to say that this is a biased way of looking at vitamin B₁₂, biased through my predilection for the making of molecules and being captivated by the ways how Nature is making them. I neglect to do justice to many other development that distinguishes B₁₂ research in the nineties so drastically from that in the seventies. The recent advent of B₁₂-on-protein X-ray structures is clearly the fulfilment of the dream of many. It undoubtedly marks the direction along which progress in B₁₂ research will tend to take place in the near future. Vitamin B₁₂, the beautiful: Excitement will not cease !

The organizers of this marvellous symposium are to be congratulated. Special thanks must go to Bernhard Kräutler, who so brilliantly hosted the symposium in his beautiful hometown Innsbruck.

Preface

This volume reviews much of the current activities in the B₁₂-field, as covered in the lectures delivered at the "4th European Symposium on Vitamin B₁₂ and B₁₂-Proteins", which was held at Innsbruck, Austria, in September 1996. This symposium had the difficult task of emulating the outstanding "3rd European Symposium on Vitamin B₁₂ and Intrinsic Factor", held in 1979 in Zürich where all the great B₁₂ names were gathered. The introductory remarks at the Zürich meeting, given by Lord Todd, looked back to the 'heyday' of B₁₂ research, celebrated in the earlier two Hamburg meetings - the heroic experiments leading to the isolation and to the structure determination by X-ray analysis. One of the highlights of the Zürich meeting was the description of the completion of the ETH-Harvard total synthesis of the vitamin by Robert B. Woodward, who sadly died only 5 months after the meeting. In the time since the meeting in Zürich, the B₁₂ community unfortunately also has lost some of its other prominent members, Lord Todd, Dorothy C. Hodgkin, Wilhelm Friedrich, Paul Dowd and Rolf Scheffold. We dedicate this volume to their memory.

Efforts aimed at the elucidation of the biosynthesis of B₁₂, another major topic at the Zürich meeting, have now been crowned by success and the complete pathway was presented in the opening key lectures in Innsbruck by Alan Battersby, Denis Thibaut and Ian Scott, and this problem can now be considered to be basically solved. Further highlights at the Innsbruck meeting dealt with the first X-ray crystal structures of two B₁₂-proteins, namely the B₁₂-binding domain of methionine synthetase from *Escherichia coli*, presented by Martha Ludwig and Rowena Matthews, and methylmalonyl-CoA mutase from *Propionibacterium shermanii*, described by Phil Evans and Peter Leadlay. Both crystal structures revealed as their most spectacular result the unexpected "base-off" mode of binding of the organometallic B₁₂-coenzymes. Many notable contributions and much stimulating discussions were centered on the mode of action of B₁₂, the third major theme of the Innsbruck meeting, and the contribution of B₁₂ to newly recognized areas of biologically important organometallic processes. Among these contributions were lectures given by the groups of Wolfgang Buckel, Steve Ragsdale, JoAnne Stubbe and Rolf Thauer. Further important work presented at Innsbruck concerned structure, reactivity and spectroscopy of B₁₂ derivatives, as well as the lecture section opened by Ebba Nexø on medical aspects of B₁₂, of B₁₂-binding proteins and of their receptors.

We have subdivided the present book into the main themes, presented at the Innsbruck symposium, i.e. biosynthesis, mode of action, structural and spectroscopic studies and clinical aspects. The manuscripts have been edited to a near common format, permitting linguistic nuances, some non-SI units and using the original figures and schemes, as supplied by the authors.

VIII Preface

We especially thank Paula Enders of the Institute of Organic Chemistry, University of Innsbruck, not only for the very substantial help in the organisation of the symposium, but also for the invaluable assistance in the preparation of the camera ready version of this book.

For the support of the meeting by the University of Innsbruck, we would like to thank in particular its Rektor, Prof. Christian Smekal, and the many helping hands and heads from the Institute of Organic Chemistry. Among them we would like to name specifically Renate Hannak, Ernst Ellmerer-Müller, Karl-Hans Ongania and Ludwig Call.

We feel confident that the exciting new facets of B₁₂ revealed at the Innsbruck meeting will stimulate experimental work to tackle the remaining fundamental questions. These concern especially the structure and function of B₁₂ dependent enzymes and their associated mechanisms, the chemistry of organometallic B₁₂ derivatives in a broader sense and the uptake, transport and role of B₁₂ in human, mammalian and microbial metabolism. And so we look forward to the 5th European Symposium on B₁₂, where the B₁₂ community will have attracted a new generation of younger and interdisciplinary researchers and at which we expect to learn about further decisive "strokes" in this area.

Innsbruck
Zürich
Newcastle

Bernhard Kräutler
Duilio Arigoni
Bernard T. Golding

September 1997

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I

B₁₂: An Overview

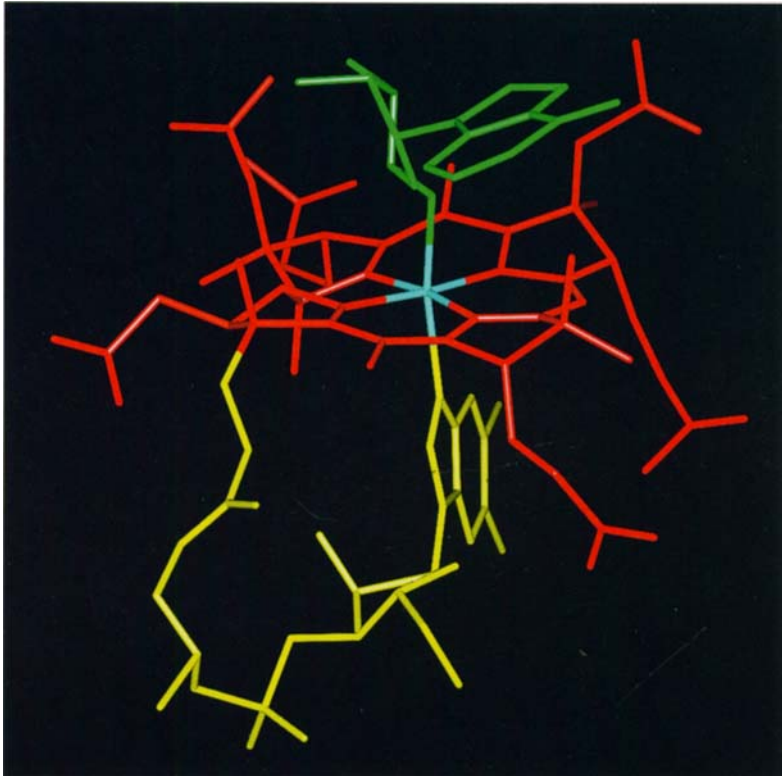


Figure 1. Model of the three-dimensional structure of coenzyme B₁₂, according to X-ray analysis

B₁₂-Coenzymes, the Central Theme

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Summary. The B₁₂-field has experienced several truly seminal developments in the last decade, with contributions from a world-wide research effort in medical, biological and chemical laboratories. Key achievements, highlighted here and described in more detail in the following chapters, center around the marvellous work leading to the detailed elucidation of the B₁₂-biosynthesis and on the pioneering X-ray analytical investigations of B₁₂-proteins, revealing unexpected structural features of protein-bound B₁₂-coenzymes. In addition, further remarkable insights into the mechanisms of B₁₂-catalyzed and related non-B₁₂ enzymatic radical reactions were gained, and novel metabolic processes were found, likely to involve B₁₂. Studies on structure, reactivity and spectroscopy continue to uncover more of the fundamental properties of B₁₂-derivatives. Investigations on B₁₂-binding proteins, their transport and interactions with their receptors in human metabolism testify to the ever increasing medical importance of B₁₂-derivatives. The B₁₂-field may be recognized as playing a central role in the newly developing, more extended area of bioorganometallic research.

1 Introduction

Close to fifty years after the first isolation of the red cobalt-complex vitamin B₁₂ (**1**) as the (extrinsic) anti-pernicious anaemia factor [1,2], the symposium on "Vitamin B₁₂ and B₁₂-Proteins" in September 1996 saw several remarkable forward strides towards the solution of some of the major B₁₂-"mysteries". To name the two major achievements, that could be reported: the elucidation of the biosynthetic pathway to B₁₂ [3-5] and the exciting first X-ray crystal structures of B₁₂-binding proteins [6,7]. As highlights of the earlier symposia in this series, held in Hamburg (1956 and 1961) [8,9] and in Zürich (1979) [10] probably the elucidation of the enigmatic structures of vitamin B₁₂ [11] and of coenzyme B₁₂ (**2**) [12] and the synthetic conquest of the vitamin B₁₂ structure [13,14] should be recalled.

The highly crystallizable, cyanide containing vitamin 1 (cyanocob(III)alamin, CNCbl) is a relatively inert Co(III)-complex and appears not to have a physiological function itself [15]. All the same, it is the most important commercially available form of the naturally occurring B₁₂-derivatives [15]. Other pharmaceutically relevant vitamin B₁₂-derivatives are the highly light sensitive and chemically more labile organometallic coenzyme forms, coenzyme B₁₂ (2, 5'-deoxy-5'-adenosylcobalamin, AdoCbl) and methylcob(III)alamin (3, MeCbl), as well as the "inorganic" B₁₂-derivatives aquocob(III)alamin (as chloride 4⁺.Cl⁻, H₂OCbl.Cl) and hydroxocob(III)alamin (5, HOCbl) (see Figure 1).

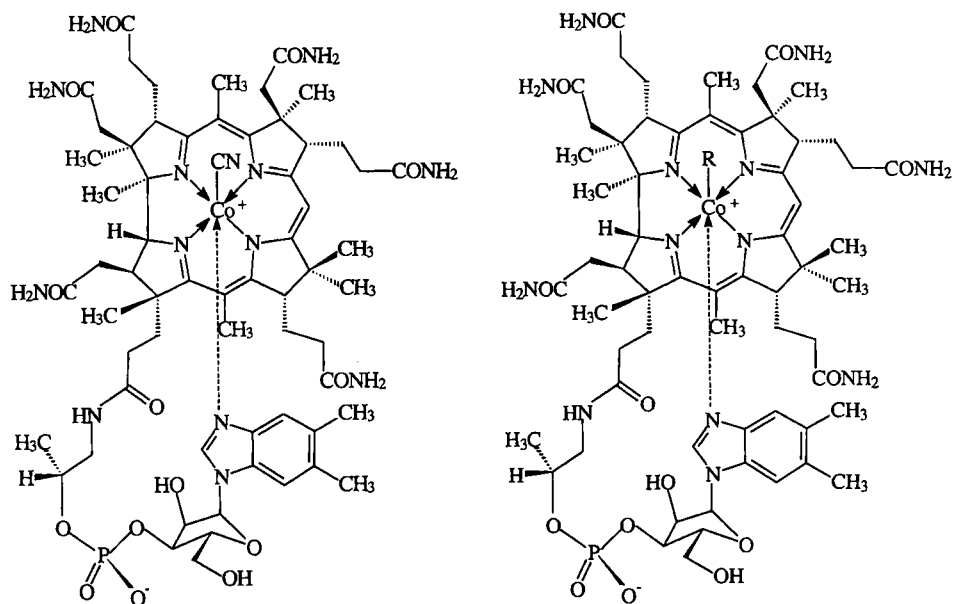
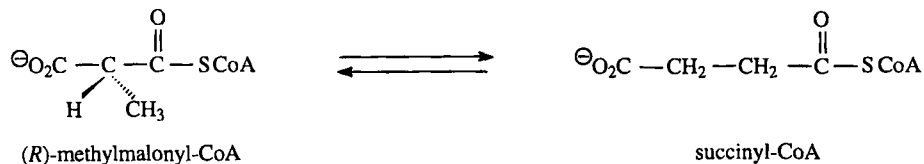


Figure 1. Structural formulae. Left: vitamin B₁₂ (1, CNCbl); right: coenzyme B₁₂ (2, R = 5'-deoxy-5'-adenosyl, AdoCbl), methylcobalamin (3, R = methyl, MeCbl), aquocobalamin cation (4⁺, R = H₂O⁺, H₂OCbl⁺), hydroxocobalamin (5, R = HO, HOCbl).

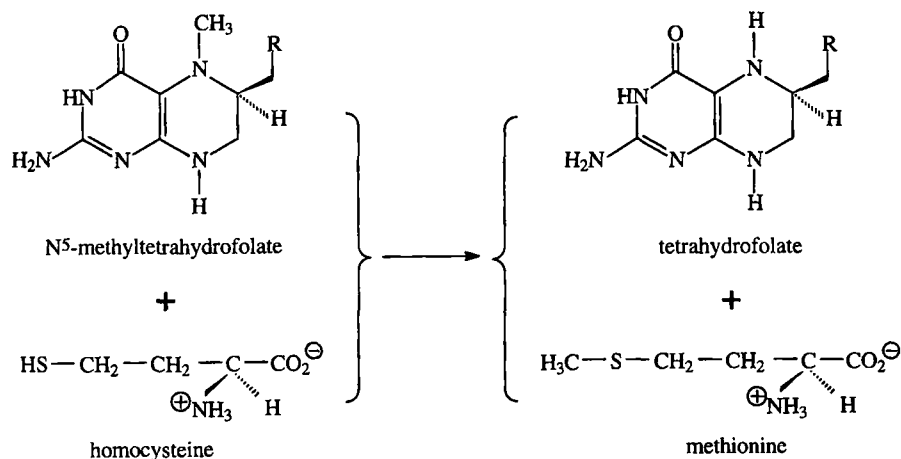
As can be gathered from many of the contributions in this book and to the now 15 year old review, "B₁₂" (edited by Dolphin) [16], the physiological roles of vitamin B₁₂-derivatives are intimately connected with their function as cofactors in enzymatic reactions. Their abilities to bind to proteins and subsequently interact with substrate molecules are questions of central importance, as are the ways in which vitamin B₁₂-forms are made available to living organisms (by their own biosynthesis [3-5,18] or else, via uptake, transport and storage [17]). The human well-being depends upon a regular supply of some of the cobalamins listed above: the B₁₂-coenzymes 2 and 3 are indispensable for human metabolism, in which the "coenzyme" 2

cocatalyzes the enzymatic rearrangement of methylmalonyl-coenzyme A to succinyl-coenzyme A [19], while methylcobalamin (3) acts as cofactor in the enzyme catalyzed methylation at the sulfur of homocysteine using a methyl group from N⁵-methyltetrahydrofolate, which leads to tetrahydrofolate and methionine [20] (see Scheme 1).

The coenzyme-B₁₂-catalyzed (*R*)-methylmalonyl-CoA/succinyl-CoA rearrangement:



The methylcobalamin-catalyzed methyl group transfer in methionine synthase:

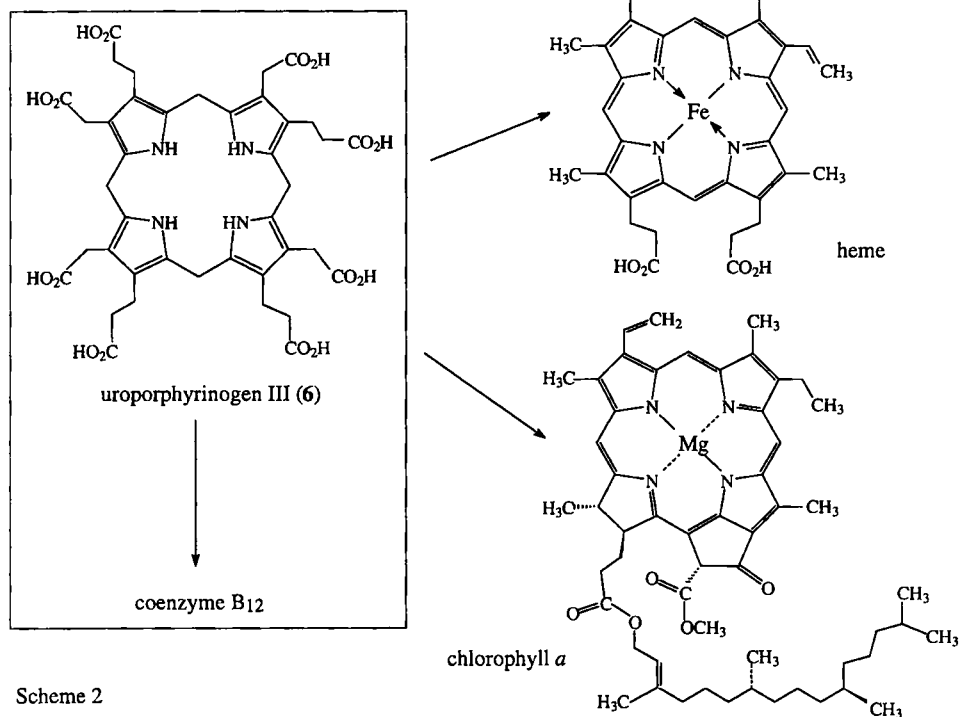


Scheme 1

The world's supply with B₁₂-derivatives depends exclusively on the activity of microorganisms, either in their natural environment, or biotechnologically exploited for the purpose of the pharmaceutically and agrotechnologically indispensable B₁₂-production. In fact, it is in the broad range of microorganisms, that B₁₂-derivatives may occupy a metabolically central position, such as in the remarkable organometallic pathway of CO₂-fixation [21]. Quite clearly, the newly developing area of bioorganometallic research nowadays still is pioneered by studies in the B₁₂-field.

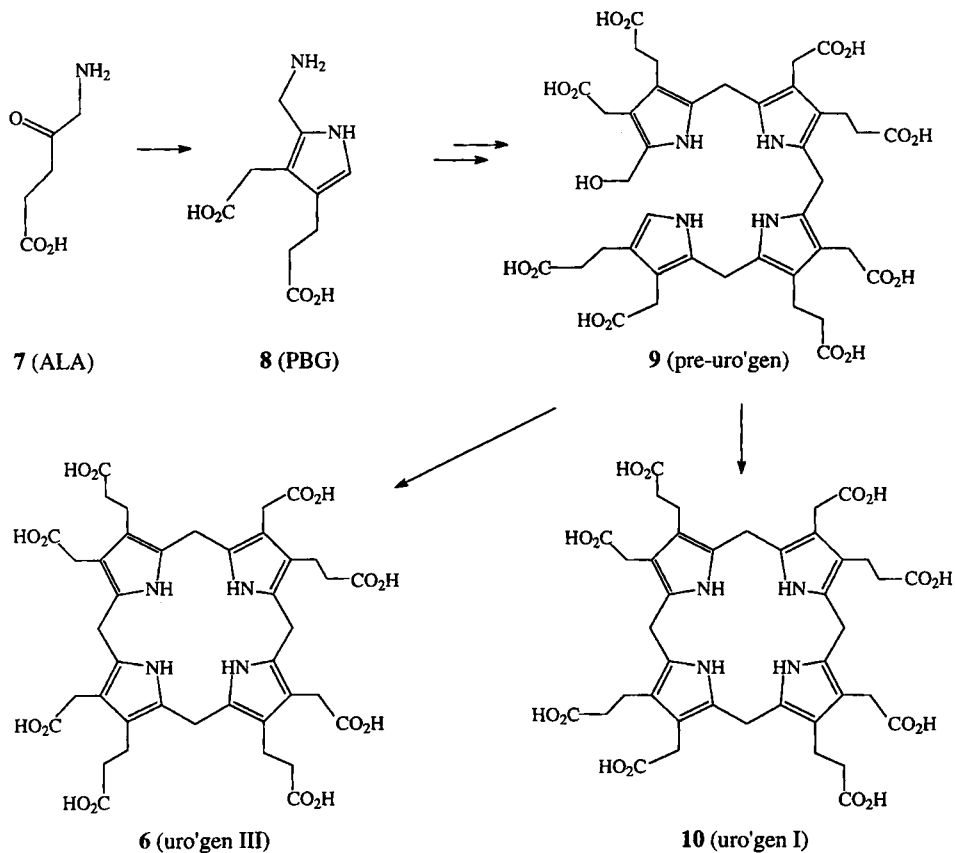
2 **B₁₂-Biosynthesis**

The total (non-enzymatic) synthesis of vitamin B₁₂ (**1**), a unique experimental and intellectual effort [13,14], was accomplished by the time of the B₁₂-Symposium in 1979 [10]. This work has not resulted in an economical method for the nonbiological synthesis of the important cobalt-complex **1**. However, together with further elegant synthetic studies on potentially biomimetic ways to the B₁₂-structure from Eschenmoser's laboratory [22], it gave insights into the chemical reactivity of relevant porphyrinoid and corrinoid complexes, and pointed to the inherent tendency of the basic moieties of the B₁₂-structure to self-assemble under (proper) nonenzymatic conditions [22]. For an era of nearly two decades, these studies provided valid abiological parallels for the earlier biosynthetic stages from uroporphyrinogen III (**6**) to the corrin ligand of B₁₂ and also gave guiding lines for investigations on the later parts of the B₁₂-biosynthesis. However, as reviewed in the reports of Battersby [3], Thibaut [4] and Scott [5] given in this book, our understanding of Nature's intricate paths from the porphyrinogen **6** to the corrin ligand of B₁₂ has undergone a spectacular deepening in these last years and the biosynthetic pathways to the B₁₂-structure, experimentally explored so far, have been shown to deviate dramatically from those considered earlier [22-24]. This is particularly so with respect to the way in which the key step of the B₁₂-biosynthesis is accomplished, a ring-contraction from a porphyrinoid to a corrinoid macrocycle.



2.1 Tetrapyrrole-Assembly

According to earlier investigations [25], Nature's path(s) to the B₁₂-structure first pass(es) through the stages common to the biosynthesis of all porphinoïd natural products (see Scheme 2). It starts with the five-carbon unit δ -aminolaevulinic acid (7, ALA), out of which a metal catalyzed enzymatic dimerization (and condensation) furnishes the functionalized pyrrole porphobilinogen (8, PBG) [25,26]. A remarkable enzyme catalyzed multistep deaminating tetrameroidization of PBG, delineated by P. M. Shoolingin-Jordan [27], then furnishes the linear tetrapyrrole pre-uroporphyrinogen (9) [25,27]. This hydroxymethylbilane would spontaneously cyclize to uroporphyrinogen I (10) [27,28], but, by rapid intervention of uroporphyrinogen III synthase, the cyclization of 9 is switched to uroporphyrinogen III (6) via a hypothetical spiro-intermediate (see Scheme 3 [27,28]). The programmed biological synthesis of 6, and not that of the more symmetrical and biosynthetically obviously less complex 10, has been proposed to be a sign of the pre-enzymatic origin of the natural porphinoïds [22, 29].

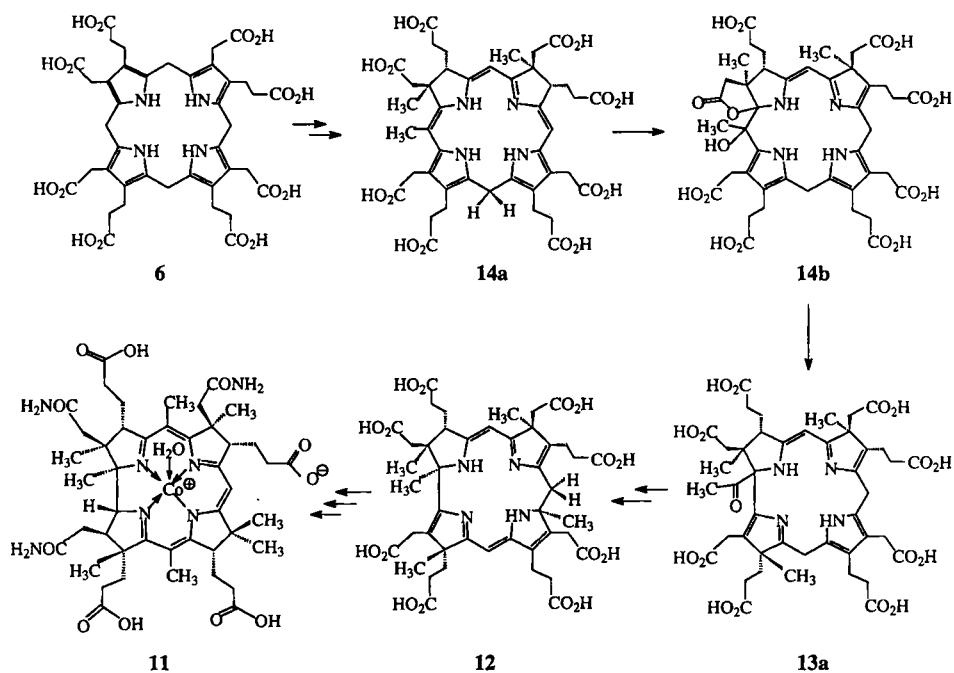


Scheme 3

2.2 Access to the Corrin Ligand

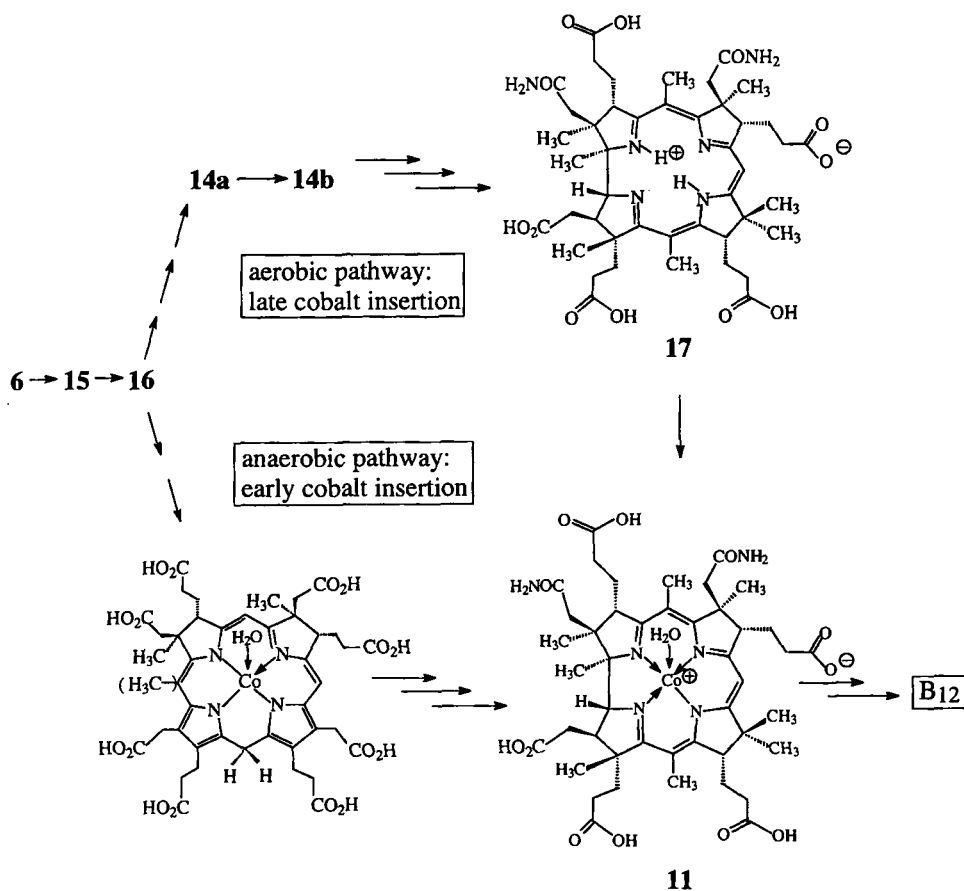
Methylation of the hexahydroporphinoid tetrapyrrole uroporphyrinogen III (**6**) with methyl groups from *S*-adenosylmethionine (SAM) at two pyrrolic β -carbons then separates off the B₁₂-biosynthesis-branch from those towards heme and the chlorophylls [25]. A total of eight methyl groups from SAM are incorporated during the sequence from uroporphyrinogen III (**6**) to the B₁₂-biosynthesis intermediate cobyrinic acid-*a,c*-diamide (**11**). One of them, which is incorporated at the crucial "western" meso-position, is lost again together with the excised meso-carbon [29] and indeed is believed to be placed there only to assist in the ring-contraction step (see below and [3-5]).

Remarkably, unlike the (oxidative) biosynthetic transformation of **6** to heme or chlorophyll, the path from **6** to the B₁₂-corrin ligand can be formulated without a formal (overall) redox change [22]. From this analysis, model considerations evolved, according to which also the individual steps of the B₁₂-biosynthesis would proceed without changes in the formal redox state of the tetrapyrroles involved [22].



Scheme 4

The detection and determination of the structure of "precorrin 6A" (**12**) [30] indicated a novel turn of the B₁₂-biosynthesis path (in the aerobic *Pseudomonas denitrificans*). The latter's unexpected "complexity" was further manifested in early 1993, when the structure of "Factor IV" (**13b**), a tetramethylated and (already) ring-contracted (corrinoid) oxidation product of "precorrin 4" (**13a**, a biosynthetic precursor of vitamin B₁₂ from *Ps. denitrificans*) [3,4,31] was reported [32]. This unanticipated structural finding and that concerning "precorrin 6A" (**12**) [3-4,30] was soon consolidated mechanistically by the isolation and determination of the structure of the enzymatic oxidation product "precorrin 3B" (**14b**, see Scheme 4 [3-5,33]). Indeed, after the structure of **13b** was revealed and from a "chemical point of view" [34], the operation of an oxidation preceding the ring-contraction step in the B₁₂-biosynthesis appeared to make sense, as it would adjust the reactivity at the crucial "western" meso-position of the hexahydrophorphinoid "precorrin 3A" (**14a**) for the incorporation of oxygen.



Scheme 5

As delineated in the following chapters by Battersby [3], Thibaut [4] and Scott [5], the biological and chemical investigations on B₁₂-biosynthesis have in fact diverged meanwhile, into studies on the aerobic *Ps. denitrificans* and (more recently also) on the anaerobic microorganism *Propionibacterium shermanii*. From this work, detailed knowledge indeed on two fairly distinct basic pathways (rather than a single one) from **6** to the highly functionalized corrinoid B₁₂-precursor cobyrinic acid-*a,c*-diamide (**11**) has evolved (see Scheme 5) [3-5]: Both paths start with the hexahydroporphinoids precorrin-1 (**15**) and precorrin-2 (**16**). The anaerobic branch then passes through the incorporation of a cobalt-ion into the porphyrinoid macrocycle, while the aerobic branch proceeds without cobalt [3-5]. The intriguing difference between the two pathways is apparently caused by the ways in which the now crucial two-electron oxidation can be achieved to prepare for the remarkable enzyme catalyzed contraction to a corrinoid. In the aerobic *Ps. denitrificans* molecular oxygen has been recognized to play the part of the oxygenating agent in this step [3-5]. In the anaerobic path, studied with *P. shermanii*, the observed early biosynthetic incorporation of the cobalt-ion now appears to be required, as the redox-active metal-ion may take up a role in mediating the crucial early two-electron oxidation of the macrocycle [5].

2.3 Completion of the B₁₂-Structure

Interestingly, the two paths of the B₁₂-biosynthesis merge at the stage of the cobyrinic acid-*a,c*-diamide (**11**), formed by cobalt-incorporation into hydrogenobyric acid-*a,c*-diamide (**17**) in the aerobic path [3,4]. The presence of the *c*-acetamide group in **17** and its tendency to reversibly form the lactam **18** by (base catalyzed) addition of the *c*-acetamide group to the corrin chromophore [35] may indeed be specifically relevant for the enzymatic metal incorporation [36]: By the reversible formation of the lactam function, the corrin chromophore would be interrupted in **18** and a kink would be introduced in the corrin ligand, by which the metal incorporation may be facilitated. This may then critically assist the enzymatic cobalt incorporation [4,36], presumed to be inherently slow with an intact corrin ligand, as judged from the notorious difficulty of chemically demetallating cobalt-corrins [37].

The later steps of the B₁₂-biosynthesis in the aerobic *Ps. denitrificans* actually pass through the stage of coenzyme B₁₂ (**2**) and require first the 5'-adenosylation of the central cobalt-ion of **11** giving Coβ-5'-adenosyl-cobyrinate-*a,c*-diamide (**19a**), followed by further amidation of side chains and the biosynthetic build-up of the nucleotide loop, via 5'-adenosylcobinamide (**19b**, AdoCbi) (see Scheme 6) [4,38]. The elucidation of the biosynthetic origin of the heterocyclic pseudo-nucleotide bases of some "complete" corrins (i.e., those, that carry an intramolecularly coordinating nucleotide function) also has revealed some unexpected turns, as reported by Renz in this book [18a] for the case of the 5',6'-dimethylbenzimidazole (DBI) base of vitamin B₁₂ (**1**).

The availability of all crucial B₁₂-biosynthetic enzymes in overexpressed form and the knowledge of their operation has recently enabled Scott and coworkers to achieve a remarkable