Biochemistry for Sport and Exercise Metabolism



WILEY



Biochemistry for Sport and Exercise Metabolism

Second Edition

Don MacLaren Liverpool John Moores University Liverpool, UK

James Morton Liverpool John Moores University Liverpool, UK



This second edition first published 2024 ©2024 John Wiley & Sons Ltd

Edition History: John Wiley & Sons Ltd (1e, 2012)

All rights reserved, including rights for text and data mining and training of artificial technologies or similar technologies. No part of this publication may be reproduced, stored in a retrieval system, or transmitted, in any form or by any means, electronic, mechanical, photocopying, recording or otherwise, except as permitted by law. Advice on how to obtain permission to reuse material from this title is available at http://www.wiley.com/go/permissions.

The right of Don MacLaren and James Morton to be identified as the authors of this work has been asserted in accordance with law.

Registered Offices

John Wiley & Sons, Inc., 111 River Street, Hoboken, NJ 07030, USA John Wiley & Sons Ltd, The Atrium, Southern Gate, Chichester, West Sussex, PO19 8SQ, UK

For details of our global editorial offices, customer services, and more information about Wiley products visit us at www.wiley.com.

Wiley also publishes its books in a variety of electronic formats and by print-on-demand. Some content that appears in standard print versions of this book may not be available in other formats.

Trademarks: Wiley and the Wiley logo are trademarks or registered trademarks of John Wiley & Sons, Inc. and/or its affiliates in the United States and other countries and may not be used without written permission. All other trademarks are the property of their respective owners. John Wiley & Sons, Inc. is not associated with any product or vendor mentioned in this book.

Limit of Liability/Disclaimer of Warranty

In view of ongoing research, equipment modifications, changes in governmental regulations, and the constant flow of information relating to the use of experimental reagents, equipment, and devices, the reader is urged to review and evaluate the information provided in the package insert or instructions for each chemical, piece of equipment, reagent, or device for, among other things, any changes in the instructions or indication of usage and for added warnings and precautions. While the publisher and authors have used their best efforts in preparing this work, they make no representations or warranties with respect to the accuracy or completeness of the contents of this work and specifically disclaim all warranties, including without limitation any implied warranties of merchantability or fitness for a particular purpose. No warranty may be created or extended by sales representatives, written sales materials or promotional statements for this work. The fact that an organization, website, or product is referred to in this work as a citation and/or potential source of further information does not mean that the publisher and authors endorse the information or services the organization, website, or product may provide or recommendations it may make. This work is sold with the understanding that the publisher is not engaged in rendering professional services. The advice and strategies contained herein may not be suitable for your situation. You should consult with a specialist where appropriate. Further, readers should be aware that websites listed in this work may have changed or disappeared between when this work was written and when it is read. Neither the publisher nor authors shall be liable for any loss of profit or any other commercial damages, including but not limited to special, incidental, consequential, or other damages.

Library of Congress Cataloging-in-Publication Data Applied for:

Paperback ISBN: 9781119605041

Cover Design: Wiley

Cover Images: © IT Tech Science/Shutterstock, © Godong/Alamy Stock Photo

Set in 9.5/12.5pt STIXTwoText by Straive, Chennai, India

Contents

Preface *xiii*

1	Energy Sources for Muscular Activity 1
1.1	Adenosine Triphosphate: The Energy Currency 2
1.2	Energy Continuum 2
1.3	Energy Supply for Muscle Contraction 3
1.4	Energy Systems and Running Speed 6
1.5	Why Can't a Marathon be Sprinted? 7
1.6	Energy Sources and Muscle 8
1.7	Can Muscle Use Protein for Energy? 9
1.8	Key Points 10
	References 11
2	Skeletal Muscle Structure and Function 13
2.1	Skeletal Muscle Structure 14
2.1.1	Gross Anatomical Structure 14
2.1.2	The Muscle Fibre 14
2.1.2.1	Sarcolemma 16
2.1.2.1	Sarcolemna 10 Sarcoplasm 17
2.1.2.2	Myofibrils 17
2.1.2.3	Sarcoplasmic Reticulum 19
2.1.2.4	Neuromuscular Junction 20
2.1.2.3	Muscle Contraction 21
2.2.1	
2.2.1	Propagation of the Action Potential 22
2.2.2	Excitation–Contraction Coupling 24 The Sliding Filament Mechanism 24
	Muscle Fibre Types 26
2.3 2.3.1	General Classification of Muscle Fibres 26
	Slow Oxidative Fibres 26
2.3.1.1	
2.3.1.2	Fast Oxidative Glycolytic Fibres 27
2.3.1.3	Fast Glycolytic Fibres 28
2.3.2	Muscle Fibre Distribution 28
2.3.3	Muscle Fibre Recruitment 29

vi	Contents	
	2.4	Muscles in Action 30
	2.4.1	Types of Muscle Contraction 30
	2.4.2	The Twitch Contraction 30
	2.4.3	The Length–Tension Relationship 31
	2.4.4	Tetanus Contractions 31
	2.4.5	Force-Velocity Relationship 32
	2.4.6	Muscle Fatigue 33
	2.5	Key Points 34
		References 35
	3	Biochemical Concepts 37
	3.1	Organization of Matter 38
	3.1.1	Matter and Elements 38
	3.1.2	Atoms and Atomic Structure 38
	3.1.3	Atomic Number and Mass Number 40
	3.1.4	Atomic Mass 40
	3.1.5	Ions, Molecules, Compounds and Macronutrients 41
	3.2	Chemical Bonding 42
	3.2.1	Ionic Bonds 42
	3.2.2	Covalent Bonds 44
	3.2.3	Molecular Formulae and Structures 44
	3.2.4	Functional Groups 46
	3.3	Chemical Reactions, ATP and Energy 46
	3.3.1	Energy 47
	3.3.2	ATP 48
	3.3.3	Units of Energy 50
	3.3.4	Types of Chemical Reactions 51
	3.3.4.1	Synthesis Reactions 51
	3.3.4.2	Decomposition Reactions 51
	3.3.4.3	Reversible Reactions 51
	3.3.4.4	Phosphorylation and Dephosphorylation Reactions 52
	3.3.4.5	Exchange Reactions 52
	3.3.4.6	Oxidation–reduction Reactions 52
	3.4	Water 53
	3.4.1	General Functions of Water 53
	3.4.2	Water as a Solvent 54
	3.5	Solutions and Concentrations 54
	3.6	Acid-Base Balance 56
	3.6.1	Acids, Bases and Salts 56
	3.6.2	pH Scale 57
	3.6.3	Buffers 58
	3.7	Cell Structure 58
	3.7.1	The Plasma Membrane 59
	3.7.2	The Nucleus 60
	3.7.3	Cytoplasm and Organelles 61

3.7.3.1	Endoplasmic Reticulum 61
3.7.3.2	Golgi Apparatus 61
3.7.3.3	Mitochondria 62
3.7.3.4	Cytoskeleton 62
3.8	Key Points 62
	References 64
4	Proteins 65
4.1	Protein Function 66
4.1.1	General Protein Function 66
4.1.1.1	Catalytic 66
4.1.1.2	Transport and Storage 67
4.1.1.3	Hormones 67
4.1.1.4	Signalling 68
4.1.1.5	Contractile 68
4.1.1.6	Structural 68
4.1.1.7	Immunological 69
4.1.1.8	Regulatory 69
4.2	Amino Acids 69
4.2.1	Amino Acid Structure 69
4.3	Protein Structure 72
4.3.1	Primary Structure 72
4.3.2	Secondary Structure 73
4.3.3	Tertiary Structure 73
4.3.4	Quaternary Structure 75
4.4	Proteins as Enzymes 75
4.4.1	Mechanisms of Enzyme Action 76
4.4.2	Factors Affecting Rates of Enzymatic Reactions 77
4.4.2.1	Substrate Concentration 77
4.4.2.2	pH 78
4.4.2.3	Temperature 78
4.4.2.4	Enzyme Concentration 79
4.4.3	Coenzymes and Cofactors 80
4.4.4	Classification of Enzymes 81
4.4.5	Regulation of Enzyme Activity 81
4.4.5.1	Covalent Modification 82
4.4.5.2	Allosteric Modification 82
4.5	Protein Turnover 83
4.5.1	Overview of Protein Turnover 83
4.5.2	DNA Structure 84
4.5.3	Transcription 86
4.5.4	The Genetic Code 86
4.5.5	Translation 88
4.6	Amino Acid Metabolism 89
4.6.1	Free Amino Acid Pool 90

viii	Contents			
	4.6.2	Transamination 91		
	4.6.3	Deamination 93		
	4.6.4	Branched Chain Amino Acids 93		
	4.6.5	Glucose-Alanine Cycle 95		
	4.6.6	Glutamine 95		
	4.6.7	The Urea Cycle 96		
	4.7	Key Points 97		
		References 98		
	5	Carbohydrates 101		
	5.1	Relevance of Carbohydrates for Sport and Exercise 101		
	5.2	Types and Structure of Carbohydrates 104		
	5.2.1	Monosaccharides 105		
	5.2.2	Disaccharides and Polysaccharides 106		
	5.2.2.1			
	5.2.2.2	Polysaccharides 109		
	5.3	Metabolism of Carbohydrates 113		
	5.3.1	Glycogenolysis 115		
	5.3.2	Glycolysis 115		
	5.3.3	Lactate Metabolism 118		
	5.3.4	The 'Link' Reaction; Production of Acetyl-CoA 118		
	5.3.5	The TCA (or Krebs) Cycle 119		
	5.3.6	Electron Transport Chain 120		
	5.3.7	Oxidative Phosphorylation 122		
	5.3.8	Calculation of ATP Generated in Glucose Oxidation 123		
	5.3.9	Overview of Glucose Oxidation 124		
	5.3.10	Fructose Metabolism 124		
	5.3.11	Gluconeogenesis 125		
	5.3.12	Glycogenesis 128		
	5.4	Key Points 129		
		References 130		
	6	Lipids 133		
	6.1	Relevance of Lipids for Sport and Exercise 133		
	6.2	Structure of Lipids 137		
	6.2.1	Classification of Lipids 137		
	6.2.2	Compound Lipids 140		
	6.2.3	Derived Lipids 141		
	6.3	Metabolism of Lipids 142		
	6.3.1	Lipolysis 142		
	6.3.2	β-Oxidation 144		
	6.3.3	Ketone Body Formation 146		
	6.3.4	Formation of Fatty Acids 147		
	6.3.5	Triglyceride Synthesis 149		

6.4	Key Points 151 References 152
7	Principles of Metabolic Regulation 155
7.1	Introduction 155
7.2	Hormones 156
7.3	Peptide Hormones, Neurotransmitters and Regulation 159
7.3.1	Adrenaline Activation of Glycogenolysis 161
7.3.2	Adrenaline Activation of Lipolysis 162
7.3.3	Insulin Activation of Glycogen Synthase 164
7.3.4	Insulin Inhibition of Lipolysis 165
7.3.5	Insulin Stimulation of Protein Synthesis 166
7.4	Steroid Hormones and Regulation 167
7.5	Allosteric Effectors 168
7.5.1	Regulation of Glycogen Phosphorylase 169
7.5.2	Regulation of PFK 169
7.5.3	Regulation of PDH 170
7.5.4	Regulation of CPT1 171
7.5.5	AMPK as a Metabolic Regulator 171
7.6	Exercise-Induced Transcription Factors 173
7.7	Regulators of Transcription 175
7.7.1	Hypoxia-Inducible Factors 176
7.7.2	Redox Balance and Sirtuins 177
7.7.3	Cell Energy Status and AMPK Signalling 177
7.7.4	Mechanical Stress, ROS and Mitogen-Activated Protein Kinase Signalling 177
7.7.5	Calcium Flux and Calcium/Calmodulin-Dependent Protein Kinase Signalling 178
7.7.6	High-Force Stimuli and Mechanosensory Signal Transduction 178
7.7.7	Regulation of Skeletal Muscle Gene Expression and Muscle Adaptation 179
7.8	Training Responses 180
7.9	Impact of Nutrition 181
7.10	Key Points 182
	References 182
8	Techniques for Exercise Metabolism 185
8.1	Introduction 185
8.2	Respiratory Analysis 186
8.3	Ergometry 189
8.4	Blood Sampling and Analysis 190
8.5	Metabolomics 192
8.6	a-v Differences 193
8.7	Muscle Biopsy 193
8.8	Nuclear Magnetic Resonance (NMR) Magnetic Resonance Spectroscopy (MRS) 195
8.9	Use of Isotopes 197

х	Contents

8.10	Key Points 199
	References 200
9	High-Intensity Exercise (HIE) 201
9.1	Overview of Energy Production and Metabolic Regulation in High-Intensity
	Exercise 201
9.1.1	Definition of High-Intensity Exercise 201
9.1.2	Energy Production During High-Intensity Exercise 202
9.1.3	Evidence of Energy Sources Used in HIE 205
9.1.4	Metabolic Regulation During High-Intensity Exercise 208
9.2	Effects of Exercise Duration 209
9.3	Effects of Nutritional Status 210
9.3.1	Can Nutritional Ergogenic Aids Help HIE? 213
9.3.1.1	Creatine 213
9.3.1.2	Alkalinizers 214
9.3.1.3	Caffeine 215
9.3.1.4	β-Alanine 216
9.4	Effects of Training 218
9.5	Mechanisms of Fatigue 222
9.5.1	Reduced ATP 222
9.5.2	Reduced PCr 225
9.5.3	Increased P _i 225
9.5.4	Lactate and H ⁺ 226
9.6	Resistance Exercise 227
9.7	Key Points 228
	References 229
10	Endurance Exercise 237
10.1	Overview of Energy Production and Metabolic Regulation in Endurance
	Exercise 238
10.1.1	Definition and Models of Endurance Exercise 238
10.1.2	Energy Production in Endurance Exercise 238
10.1.3	Overview of Metabolic Regulation in Endurance Exercise 239
10.2	Effects of Exercise Intensity 240
10.2.1	CHO Metabolism 241
10.2.1.1	Muscle Glycogenolysis 241
10.2.1.2	Plasma Glucose Utilization, Muscle Glucose Uptake and Glycolysis 243
10.2.1.3	Carbohydrate Oxidation 243
10.2.2	Lipid Metabolism 243
10.2.2.1	Adipose Tissue Lipolysis and FFA Availability/Delivery 244
10.2.2.2	FFA Transport into the Cytosol 246
10.2.2.3	FFA Transport Across Mitochondrial Membranes 246
10.2.2.4	Does Malonyl-CoA Regulate LCFA Uptake? 247
10.2.2.5	Does Free Carnitine Availability Regulate LCFA Uptake? 248
10.2.2.6	Does Exercise-Induced Decreases in Muscle pH Reduce CPTI Activity? 248

10.2.2.7	So-called Fat _{max} 248
10.3	Effects of Exercise Duration 250
10.4	Effects of Nutritional Status 252
10.4.1	CHO-Loading and Muscle Glycogen Availability 252
10.4.2	Fat-Loading Strategies 254
10.4.3	Pre-exercise and During-Exercise CHO Ingestion 257
10.4.4	Pre-exercise FFA Availability 262
10.5	Effects of Training Status 264
10.5.1	CHO Metabolism 265
10.5.2	Lipid Metabolism 267
10.5.3	Protein Metabolism 270
10.6	Mechanisms of Fatigue 272
10.7	Key Points 275
	References 277
11	High-intensity Intermittent Exercise 283
11.1	Overview of Energy Production in Intermittent Exercise 283
11.1.1	Definition and Models of Intermittent Exercise 283
11.1.2	Energy Systems Utilized in Intermittent Exercise 284
11.2	Metabolic Regulation in Intermittent Exercise 285
11.3	Effects of Manipulating Work–Rest Intensity and Ratio 292
11.4	Effects of Nutritional Status 296
11.4.1	Muscle Glycogen Availability 296
11.4.2	Pre-exercise CHO Ingestion 297
11.4.3	CHO Ingestion During Exercise 298
11.5	Muscle Adaptations to Interval Training 301
11.6	Mechanisms of Fatigue 308
11.6.1	Carbohydrate Availability 308
11.6.2	PCr Depletion 310
11.6.3	Acidosis 311
11.6.3.1	Lactate 312
11.6.3.2	Reduced pH 313
11.6.3.3	Lactate and H ⁺ Transport 314
11.6.4	Extracellular Potassium 315
11.6.5	Reactive Oxygen Species (ROS) 317
11.6.6	P _i Accumulation and Impaired Ca ²⁺ Release 318
11.7	Key Points 319
	References 320

Index *327*

Preface

My goodness, it has been 13 years since we published the first edition of this text - in essence, based on lectures we provide for our undergraduate and postgraduate sport and exercise science and the sports/performance nutrition students at Liverpool John Moores University. Guess it is time for an update! The world does not stand still and certainly, neither does scientific endeavour. As Isaac Newton stated, 'To me there has never been a higher source of earthly honour or distinction than that connected with advances in science'. This is so true of the significant developments in techniques over the past 13 years or so – notably in the understanding of not only cellular and molecular control mechanisms but also cell structures. It is the technical developments that have enabled scientists to get a greater understanding of research into human metabolism. Tools such as magnetic resonance imaging and spectroscopy, mass spectrometry, more sophisticated ergometers and improvements in some biochemical testing apparatuses have contributed to the plethora of scientific investigations reported in peer-reviewed journals. To this end, we have updated our original text to that which you are ready to explore. Of course, there is a limit to what we can change since most of the biochemistry is in essence the same. To this end, we have made some subtle modifications. We have included a chapter on biochemical techniques in order to furnish you with some background as to how the reported metabolic changes during exercise have been achieved and have made a few additions in the three chapters on metabolism.

As sport and exercise scientists, ultimately, we are left with the task of critically evaluating research (both our own and those of others) concerning exercise metabolism in the hope of integrating training and nutritional strategies which maximize performance. Underpinning all of this demands a knowledge and understanding of some biochemical principles associated with the macronutrients involved in energy production and the likely mechanisms controlling these events. We hope that after completing this text of *Biochemistry for Sport and Exercise Metabolism*, you will possess the appropriate platform for which to do so! Furthermore, we hope you will have been stimulated sufficiently to further engage in a deeper understanding of your chosen area of interest. Best wishes and good luck!

Don MacLaren and James Morton Research Institute for Sport & Exercise Sciences Liverpool John Moores University

1

Energy Sources for Muscular Activity

Learning Outcomes

After studying this chapter, you should be able to:

- outline the key energy sources for exercise;
- distinguish between anaerobic and aerobic sources of energy;
- describe the essential structure of ATP;
- draw and explain the components of the energy continuum;
- describe the role of PCr in ATP synthesis;
- explain how PCr is resynthesized;
- describe the involvement of carbohydrates and fats as energy sources for exercise;
- explain reasons why an athlete is unable to sprint a marathon;
- describe the amounts and sources of energy in the body and their rates of energy formation:
- show how and where the main energy sources are derived and utilized;
- discuss how amino acids can be used as an energy source during exercise.

This chapter presents a brief overview of the energy sources used by muscles in order to engage in various activities. It is a 'taster' that will (hopefully) encourage you to delve a bit more deeply into the basic biochemistry of the macronutrients which provide energy, as well as to gain an understanding of the likely regulation of the processes which produce energy. From this perspective, this chapter examines the energy-yielding processes from a superficial level in addressing issues of energy for sprinting and for more prolonged events.

Key words

energy continuum aerobic energy sources protein synthesis energy sources for exercise anaerobic energy sources protein degradation

1.1 Adenosine Triphosphate: The Energy Currency

In order for muscles to contract and provide movement, energy is required. Such energy is provided by adenosine triphosphate (ATP) and is the only energy capable of being used for muscle contraction in humans. Figure 1.1 provides the structure of an ATP molecule. As you can see from this diagram, ATP consists of a base (adenine) attached to a sugar (ribose), to which is attached three phosphate molecules. The phosphates are attached by 'high energy' bonds which, when removed, provide energy.

$$ATP \leftrightarrow ADP + P_i + Energy$$
 (7.3 kcal or 30.5 kJ)

The process is reversible, which means that ATP may be re-formed from adenosine diphosphate (ADP) as long as there is sufficient energy to restore the missing phosphate molecule on to the ADP. The latter can be achieved by phosphocreatine (PCr) or by processes such as anaerobic glycolysis, and aerobic processes.

The stores of ATP in muscle tissue are rather limited, so there is a constant need to resynthesize it for survival, let alone movement. The amount of ATP in a muscle cell amounts to 25 mM/kg dry muscle or about 40–50 g in total, which is sufficient to enable high intense activity for around 2–4 seconds if it is the only useable source of energy available. This is not a great amount – hence the importance of resynthesis of ATP at rates sufficient to enable appropriate levels of exercise to ensue, i.e. fast rates of resynthesis for sprinting and slower rates for prolonged exercise.

1.2 Energy Continuum

The major energy sources for exercise are dependent on the intensity and duration of the activity. Examination of Figure 1.2 highlights that there appears to be three such sources, i.e. PCr, glycolytic and aerobic. These energy-producing processes predominate exercise from 1 to 10 seconds, 10 to 60 seconds and beyond 60 seconds respectively.

Another way of expressing the energy continuum is represented in Figure 1.3, which shows the major energy sources for running events of varying distances. Note that short, highly intense sprinting bouts lasting 1–10 seconds use PCr predominantly, while events such as the 400 m mainly use anaerobic glycolysis, and thereafter aerobic metabolism predominates.

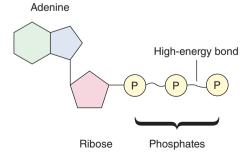


Figure 1.1 Adenosine triphosphate (ATP).

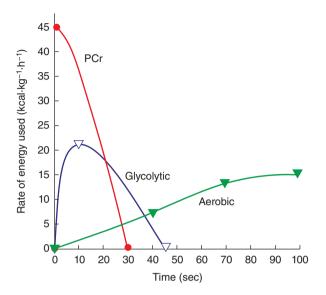
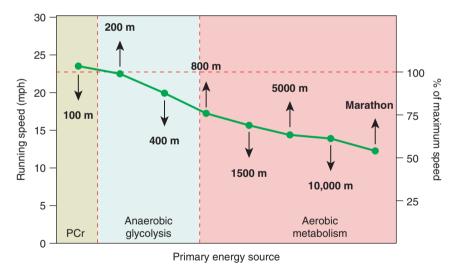


Figure 1.2 Energy continuum.



Primary energy sources for different running distances.

1.3 **Energy Supply for Muscle Contraction**

ATP is not stored to a great degree in muscle cells. Therefore, once muscle contraction starts, the regeneration of ATP must occur rapidly. There are three primary sources of ATP; these, in order of their utilization, are PCr, anaerobic glycolysis and aerobic processes.

Energy from ATP derives from cleaving the terminal phosphate of the ATP molecule. The resulting molecule is ADP. PCr converts ADP back to ATP by donating its phosphate in the presence of the enzyme creatine kinase (CK), and in turn the PCr forms creatine (Cr), i.e. the dephosphorylated form of PCr.

$$ADP + PCr \underset{CK}{\Leftrightarrow} ATP + Cr$$

The reaction of PCr with ADP to form ATP is very rapid, but is short-lived since the cell does not store high amounts of PCr (the muscle concentration of PCr is about 80 mM/kg dry muscle or 120 g in total). However, during short, high-intensity contractions, PCr serves as the major source of energy. This form of energy generation is sometimes referred to as an aerobic alactic, because it neither produces lactic acid nor requires oxygen. It is of paramount importance in sports requiring bursts of speed or power, such as sprints of 1 to10 seconds, lifting weights, engaging in a high/long jump or a throw in an athletics field event.

Figure 1.4 provides a schematic to show the synthesis of ATP from ADP using PCr at the muscle crossbridge, as well as the regeneration of PCr from Cr by ATP at the mitochondria. This is known as the 'PCr shuttle'.

Thus, Cr is produced from PCr during intense bouts of exercise, while Cr is re-phosphorylated to PCr by ATP produced in the mitochondria during an aerobic recovery phase. Oxygen is needed for recovery of PCr, as can be seen in Figure 1.5, which clearly demonstrates that recovery of exercise-depleted PCr only happens when the blood supply to the exercising muscle is not occluded, i.e. there is an intact blood supply taking oxygen to the cells. If the blood supply is occluded (e.g. via a tourniquet), then PCr resynthesis fails. As a consequence, there is the need for a low level (so-called active) recovery in between bouts of intense exercise.

The enzyme CK, which regulates PCr activity, exists in a number of forms known as isoforms (this will be dealt with later). Note that not only is there a CK which favours the formation of ATP from PCr, but there is also another form, CK_{mito} , which is present at the mitochondria and favours the synthesis of PCr from Cr using ATP. In effect, the same enzyme (CK) but in different isoforms which results in either the breakdown or synthesis of PCr.

You should also note from Figure 1.5 that there is a rapid loss of PCr during intense exercise and that it is rapidly recovered (PCr stores may even be depleted if the exercise is sufficiently intense or prolonged). Nearly 75% of PCr is resynthesized within the first

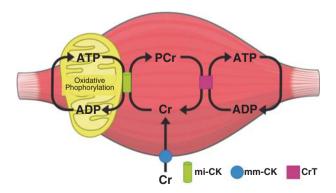


Figure 1.4 PCr shuttle (mi-CK is mitochondrial CK; mm-CK is skeletal muscle CK; CrT is creatine transporter).

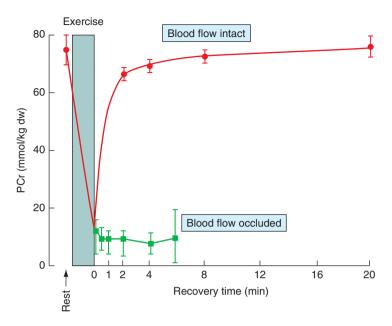


Figure 1.5 Resynthesis of PCr after exercise with and without an occluded blood supply (adapted from Hultman *et al.*, 1990).

minute of recovery and the rest over the next 3–5 minutes. The graph is biphasic, i.e. rapid restoration at first, then a second, slower phase.

As soon as muscle contraction starts, the process of anaerobic glycolysis also begins. Anaerobic glycolysis does not contribute as large an amount of energy as PCr in the short term, but its contribution is likely to predominate from 10 to 60 seconds.

During glycolysis, locally stored muscle glycogen, and possibly some blood-borne glucose, supply the substrate for energy generation. Glycolysis takes place in the cytoplasm, where no oxygen is required, so the process is called anaerobic. It has been referred to as 'anaerobic lactic', since lactic acid is formed as the end product. Sufficient lactic acid formation can lower the pH of the cell (i.e. make it more acid) to the extent that further energy production may be reduced.

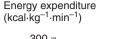
The major substrate for anaerobic glycolysis (see equation below) is glycogen stored within the muscle, so prior hard exercise without adequate repletion of glycogen will limit further high-intensity short-term work.

Glycogen
$$\rightarrow$$
 Glucose-1-P \rightarrow Lactic acid + ATP

Exercise beyond 60 seconds requires mainly aerobic energy sources, such as the complete oxidation of glucose or fatty acids to carbon dioxide and water. These processes necessitate oxygen and take place in the mitochondria of the cells. The equations below illustrate the essence of aerobic metabolic reactions:

Glucose + Oxygen → Carbon dioxide + Water + ATP Fatty acid + Oxygen → Carbon dioxide + Water + ATP

More detail about these processes are presented in Chapters 5 and 6.



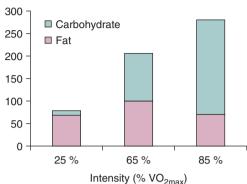


Figure 1.6 Carbohydrate and fat use at three exercise intensities (adapted from Romijn *et al.*, 1993).

Aerobic activities invariably occur at lower exercise intensities (which are those lasting longer than one minute), and the contributions of carbohydrate and fat at these levels of intensity can be realized in Figure 1.6. Note that fats contribute a greater percentage (and amount) of energy at 25% VO_{2max} (i.e. walking pace) than carbohydrate, around 50% of the energy at 65% VO_{2max} (i.e. steady state pace), and around 25% of the energy at 85% VO_{2max} (i.e. an intense aerobic bout with some significant anaerobic energy involved).

1.4 Energy Systems and Running Speed

Based on world record times, humans can maintain maximum sprinting speed for approximately 200 m. The average speeds for the 100 and 200 m world records are similar, at 22.4 and 21.6 mph, respectively. However, with increasing distances, average speeds decline. The average speed for the marathon world record is about 12 mph, which is 55% of the world sprint record. This is quite a remarkable pace, since the marathon distance is more than 200 times the length of a 200 m race.

Although natural selection plays a crucial role in elite sprinting and marathon performance, the energy systems must also be highly trained and exercise-specific to be successful. For example, the energy needed to maintain an average sprinting speed of 22 mph for 200 m or less, and that required for an average running speed of 12 mph for the marathon, are acquired by two very different systems (the predominant energy systems required for running at different speeds can be seen in Figure 1.3). The primary energy source for sprinting distances up to 100 m is PCr. From 100 to 400 m, anaerobic glycolysis is the primary energy source. For distances longer than 800 m, athletes rely primarily on aerobic metabolism.

The rate of glycogen and fat utilization varies according to the relative running speed. Although the rate of glycogen utilization is low while running a marathon, the duration of the event increases the possibility of depleting glycogen stores. In contrast, the rate of glycogen utilization is substantially higher during a 5000 m run, but glycogen depletion is not a concern because of the shorter duration of the event.

Maximum maintainable speed decreases by approximately 7 mph as running distance increases from 200 to 1500 m. However, as the distance increases from 1500 m to 42.2 km,

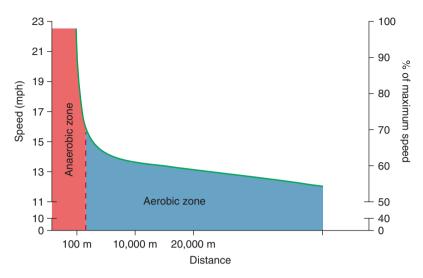


Figure 1.7 Sustainable running speed and distance run.

maximum maintainable speed only drops by an additional 3.5 mph. On average, a healthy, fit, non-elite, male athlete can be expected to sprint at an average speed of 16–18 mph for 100–200 m and at approximately 6–8 mph for a marathon (see Figure 1.7).

1.5 Why Can't a Marathon be Sprinted?

Figure 1.7 clearly demonstrates the inability to sustain high running velocities for a protracted duration. So why is an athlete unable to keep up higher running speeds over a marathon distance? The different energy sources have already been noted above, but what it is necessary to understand is that each of these energy sources resynthesizes ATP at varying rates. Table 1.1 highlights the likely rates of ATP production, and you should note the hierarchy.

The PCr system is the most rapid of these ATP-producing systems. A calculated rate of $9 \text{ mM} \cdot \text{kg}^{-1} \cdot \text{s}^{-1}$ dry muscle is more than twice as fast as ATP generation from anaerobic glycolysis which in turn is twice as fast as aerobic oxidation of carbohydrates. Furthermore,

Metabolic process	Maximum power (mM⋅kg ⁻¹ ⋅s ⁻¹)		
PCr → ATP	9		
$CHO \rightarrow lactate + ATP$	4		
$CHO \rightarrow CO_2 + H_2O + ATP$	2		
$Fat \rightarrow CO_2 + H_2O + ATP$	1		

Table 1.1 Maximum rates of energy production.

the aerobic breakdown of carbohydrates produces ATP at twice the rate of fats (i.e. 2 vs. $1 \text{ mM} \cdot \text{kg}^{-1} \cdot \text{s}^{-1}$). It thus seems that energy processes in the cytoplasm produce ATP at a faster rate than those which require oxidation via the mitochondria, and that carbohydrates produce ATP quicker than fats.

In Chapters 5 and 6, we will see that whereas PCr generation of ATP is a single reaction, anaerobic glycolysis entails 10 reactions, aerobic breakdown of glucose necessitates around 26 reactions (if the TCA cycle is used twice), and somewhere in the region of 90–100 reactions are required for complete fatty acid oxidation. No wonder, there are varying rates for ATP production.

Since the muscle stores of PCr are rather limited, and the end product of the rapid ATP generation from anaerobic glycolysis produces lactic acid, it would appear that it is not possible for an athlete to keep running at a sprint pace when undertaking a marathon – either they would run out of PCr, or the pH of their muscles would be significantly reduced due to lactic acid production. In addition, there are also limited stores of muscle and liver carbohydrate (glycogen) which would seem to be problematic as a source of energy for a complete marathon, so the need to employ fatty acids is important in energy production. Fatty acids produce the slowest rates of ATP synthesis – hence the fact that when these stores are engaged, running speeds are lowered.

1.6 Energy Sources and Muscle

Figure 1.8 provides an important overall schema as to how carbohydrates and lipids contribute as the major energy sources for muscle activity – the relevance of amino acids will

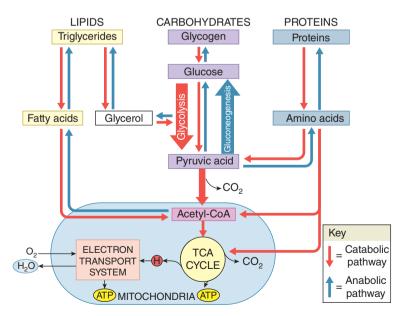


Figure 1.8 Schema of key sources and processes for skeletal muscle to derive energy during exercise.

	ATP	PCr	Anaerobic glycolysis	Carbohydrate oxidation	Fatty acid oxidation
Total amount	40 g	120 g	350 g of CHO	500 g of CHO	15,000 g of fatty acids
Duration of exercise before depleted stores		4–6 sec	1–2 min	1-2 h	>6 h
Max rate of ATP synthesis $(mM \cdot kg^{-1} \cdot s^{-1})$		9	4	2	1

Table 1.2 Energy sources available to working muscle including amounts and likely duration before depletion.

be described in Chapter 4. It is a figure that we shall return to, albeit in greater detail, throughout the text. In summary, you will note that the so-called anaerobic energy processes from ATP, PCr and anaerobic glycolysis occur in the cytoplasm, whereas the aerobic energy processes from muscle glycogen, blood-borne glucose, intramuscular triacylglycerol and blood-borne free fatty acids take place in the mitochondria. The blood-borne glucose arises from the liver due to both the breakdown of liver glycogen and the ability of the liver to produce glucose from glycerol (in essence the breakdown of triacylglycerol) as well as amino acids such as leucine and glutamine. Furthermore, it is clear that in order for blood-borne sources to enter a muscle there is a need for some form of membrane transport mechanism for each of the extramuscular energy sources to enter a muscle cell as rapidly as possible. In Chapters 5 and 6 we will explore, in more detail, the nature of these transporters as well as examine the production of energy in skeletal muscle.

A final consideration is to deliberate as to how much energy these sources contain and how quickly a muscle can utilize them. Table 1.2 highlights a number of key points in relation to energy sources for muscle. These include:

- 1. the total amount of the energy source, from which it is quite apparent that the faster ATP-producing sources normally present within the muscle are limited (notably PCr and intramuscular glycogen for anaerobic glycolysis) whereas the slower ATP-producing and essentially extramuscular sources are more plentiful;
- 2. the likely duration for which these energy sources will last if they are the only source of ATP production, from which it is apparent that extramuscular sources provide longer lasting energy compared with intramuscular sources;
- 3. the maximal rates by which they can produce ATP from which it is apparent that intramuscular sources which do not require oxygen (anaerobic processes) rapidly generate ATP, whereas those requiring oxygen (aerobic processes) generate ATP more slowly.

Can Muscle Use Protein for Energy? 1.7

So far there has been little mention of using proteins for energy. Muscles are made of proteins in the main, but can muscle protein provide energy? The answer is, to a limited extent,

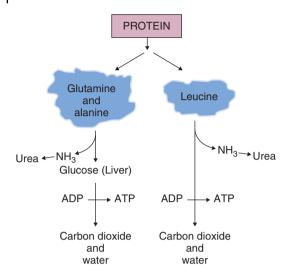


Figure 1.9 Likely use of amino acids for energy.

yes. A major difference between carbohydrates and fats is that they are essentially made up of carbon, hydrogen and oxygen only, whereas protein molecules also contain an amino group (i.e. a nitrogen). The end result of carbohydrate and fat oxidation is the generation of carbon dioxide and water, whereas oxidation of proteins requires the removal of nitrogen.

Figure 1.9 illustrates the fact that amino acids (the basic structural component of proteins) can, after the removal of the nitrogen (which ends up as urea), be converted to carbohydrates, which can then be oxidized. During prolonged exercise, the amino acids alanine and glutamine are converted to glucose in the liver, and the glucose is then oxidized by muscle. In addition, the muscle also has a limited capacity to oxidize the amino acid leucine. In total, amino acids usually accounts for 5% of the energy needed by muscle.

1.8 Key Points

- Adenosine triphosphate (ATP) is the useable form of energy for muscle contraction.
- Phosphocreatine (PCr), anaerobic glycolysis and aerobic processes enable ATP to be resynthesized during exercise.
- High intensity bouts of exercise demand a faster rate of ATP generation if the activity is
 to proceed and this is achieved by the faster 'anaerobic' sources, i.e. PCr and anaerobic
 glycolysis.
- Low to moderate bouts of exercise use aerobic energy processes such as complete oxidation of carbohydrates and fats.
- ATP and PCr content are limited in muscle and hence the reduced capability to engage in very intense levels of activity for prolonged periods.
- Anaerobic glycolysis results in lactic acid formation which is considered by some research to contribute to fatigue.
- Aerobic energy sources can be present within muscle (intramuscular glycogen and intramuscular triacyglycerol) or brought to the muscle as exogenous sources (glucose, acids and amino acids).

- Carbohydrate sources of energy (glycogen) are limited in comparison with fat sources.
- Amino acids from protein breakdown can contribute to energy production in a limited manner.

References

Hultman, E., Bergstrom, M., Spriet, L.L. & Soderlund, K (1990) Energy metabolism and fatigue. Biochemistry of Exercise VII 21, 73-92.

Romijn, J.A., Sidossis, L.S., Gastaldelli, A., Horowitz, J.F., Endert, E. & Wolfe, R.R. (1993) Regulation of endogenous fat and carbohydrate metabolism in relation to exercise intensity and duration. American Journal of Physiology 265, E380-E391.

7

Skeletal Muscle Structure and Function

Learning Outcomes

After studying this chapter, you should be able to:

- describe the gross anatomical structure of skeletal muscle;
- list the main sub-cellular components of the muscle fibre and outline their location and function;
- draw and label the sarcomere including the A-band, I-band, M-line and H-zone;
- describe the structure of the thick and thin filaments;
- define the term motor unit;
- explain the structure and function of the neuromuscular junction;
- explain and outline the main stages involved in the process of muscle contraction;
- compare and contrast the structural, biochemical and functional properties of type I, type IIa and type IIx muscle fibres;
- explain how muscle fibres are recruited with varying exercise intensities;
- define what is meant by lengthening, shortening and isometric muscle contractions;
- highlight and explain the phases of a twitch contraction;
- describe how stimulation frequency affects contractile force and define the term tetanus;
- explain the length-tension and force-velocity relationships;
- define the terms fatigue, central fatigue and peripheral fatigue.

Key words

skeletal muscle structure muscle relaxation muscle fibre types muscle contraction sliding filament mechanism central and peripheral fatigue

Skeletal Muscle Structure 2.1

In Chapter 1, we provided an introductory overview of the energy sources and systems involved in producing energy for muscular activity. Given that our focus is on the provision of energy for working skeletal muscle, it follows that we should now develop a sound understanding of both the structure and function of muscle itself. These topics are therefore the central theme of this chapter. Much of what will be discussed herein relates to the combined interplay between a variety of sub-cellular components and proteins involved in co-ordinating muscle contraction. For this reason, readers not familiar with general cell structure or protein function may wish to initially read sections of Chapter 4 prior to reading this chapter.

Skeletal muscle can be considered an organ as it is composed of cells from multiple tissues, i.e. nervous tissue, connective tissue, etc. and, of course, cells from muscle tissue itself. In this context, skeletal muscle is the largest organ in the human body, comprising 40-50% of total body weight.

There are over 600 muscles in our bodies, all performing common functions:

- 1. producing body movements;
- 2. maintaining posture;
- 3. storing and moving substances within the body; and
- 4. generating heat.

Skeletal muscle is so called because it primarily functions to move bones of the skeleton and as such, muscle tissue is connected to bones by connective tissue known as tendons. Each end of a specific skeletal muscle is attached to a bone that is essentially stationary (termed the point of origin) or to a bone that is moved (termed the point of insertion) during the specific muscle contraction. For example, the biceps brachii muscle has its points of origin and insertion in the scapula and radius bones, respectively.

2.1.1 Gross Anatomical Structure

The gross anatomical structure of skeletal muscle is shown in Figure 2.1. Surrounding the entire whole muscle is a strong sheet of fibrous connective tissue known as fascia. Three separate layers of connective tissue then extend from the outermost layer of fascia to strengthen and protect the muscle further.

The initial layer encompassing the whole muscle is known as the *epimysium*. If we were to cut through the epimysium, we would then encounter the perimysium. The perimysium encloses groups of 10-100 muscle cells and essentially separates them into muscle cell bundles known as fascicles. In turn, each muscle cell within this bundle is also separated from one another by a layer of connective tissue known as the endomysium.

The Muscle Fibre 2.1.2

Skeletal muscle cells are more commonly referred to as muscle fibres and they differ from many other cells in the body for a number of reasons. Muscle fibres have a unique shape